metal-organic papers

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Key indicators

Single-crystal X-ray study T = 200 KMean σ (C–C) = 0.007 Å H-atom completeness 84% Disorder in solvent or counterion R factor = 0.032 wR factor = 0.036 Data-to-parameter ratio = 13.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

cis,cis,cis-Aquabis[bis(diphenylphosphino)methane- $\kappa^2 P, P'$]chlororuthenium(II) hexafluorophosphate methanol 1.73-solvate

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The title compound, $[RuCl(C_{25}H_{22}P_2)_2(H_2O)]PF_6\cdot1.73CH_4O$, contains a pseudo-octahedral ruthenium complex cation with two bidentate diphosphine ligands and *cis*-disposed chloro and aqua ligands, together with a disordered hexafluorophosphate anion and disordered methanol solvent molecules. Identification as an aquaruthenium(II) complex rather than a hydroxoruthenium(III) complex was confirmed by cyclic voltammetry.

Comment

For the most part, the bond distances and angles of the title complex, (I), are unremarkable and similar to those of related complexes (Hartwig *et al.*, 1991). Deviations from octahedral geometry at the Ru atom result from the presence of two chelating bis(diphenylphosphino)methane ligands [P1-Ru1-P2 = 71.08 (3)° and P3-Ru1-P4 = 71.33 (3)°]. The Ru-P distances for the mutually *trans* P-donor atoms [Ru1-P1 = 2.345 (1) Å and Ru1-P3 = 2.3839 (9) Å] are significantly longer than those *trans* to the chloro and aqua ligands [Ru1-P2 = 2.3028 (8) Å and Ru1-P4 = 2.312 (1) Å].



Experimental

A solution of *cis*-[RuCl₂(dppm)₂] (50 mg, 0.053 mmol), sodium hexafluorophosphate (15 mg, 0.089 mmol), water (2 ml) and triethylamine (2 ml) in dichloromethane was stirred for 16 h. The solvent was removed under reduced pressure and the title complex was purified by passing a dichloromethane extract of the residue through an alumina plug with dichloromethane. Compound (I) was then recrystallized from dichloromethane/methanol, affording green crystals (yield 42.5 mg, 75%). Analysis calculated for C₅₀H₄₆ClF₆O-P₅Ru·1.73CH₄O: C 55.34, H 4.65%; found: C 55.39, H 4.28%. IR (CH₂Cl₂): ν PF₆ 846 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ 6.35–8.20 (*m*, 40, Ph), 5.30 (*m*, 4, CH₂). UV–vis: λ (THF) 346 nm, ε 600 M⁻¹ cm⁻¹. E-chem: Ru^{II/III} = 0.51 V, Ru^{III/IV} = 1.29 V.

Crystal data

$[RuCl(C_{25}H_{22}P_2)_2(H_2O)]PF_6$	$D_x = 1.474 \text{ Mg m}^{-3}$
1.73CH ₄ O	Mo $K\alpha$ radiation
$M_r = 1123.67$	Cell parameters from 50 753
Monoclinic, Cc	reflections
a = 20.4427 (2) Å	$\theta = 3-27^{\circ}$
b = 12.7593 (2) Å	$\mu = 0.58 \text{ mm}^{-1}$
c = 21.0262 (2) Å	$T = 200 { m K}$
$\beta = 112.6186 \ (7)^{\circ}$	Block, green
$V = 5062.53 (11) \text{ Å}^3$	$0.30 \times 0.26 \times 0.17 \text{ mm}$
Z = 4	

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C22

C2

 P^{2}

01

Ru

C26

C25

C50

C48

P3

CIÌ

C47

C10

C49

C40

C11

C12

Data collection

Nonius KappaCCD diffractometer

 φ and ω scans Absorption correction: by integration [Coppens (1970); implemented in maXus (Mackay et al., 2000)] $T_{min} = 0.883, T_{max} = 0.957$ 50 753 measured reflections

Refinement

Refinement on F R = 0.032 wR = 0.036 S = 1.058523 reflections 637 parameters H-atom parameters constrained Chebychev polynomial with five parameters: 0.771, 0.341, 0.66,

Table 1

Selected geometric parameters (Å, °).

Ru1-Cl1	2.4656 (8)	P2-C13	1.818 (3)
Ru1-P1	2.345 (1)	P2-C19	1.836 (4)
Ru1-P2	2.3028 (8)	P2-C49	1.857 (3)
Ru1-P3	2.3839 (9)	P3-C25	1.839 (3)
Ru1-P4	2.312(1)	P3-C31	1.820 (3)
Ru1-O1	2.196 (3)	P3-C50	1.830 (4)
P1-C1	1.816 (4)	P4-C37	1.827 (4)
P1-C7	1.827 (4)	P4-C43	1.824 (4)
P1-C49	1.830 (3)	P4-C50	1.836 (3)
Cl1-Ru1-P1	92.74 (3)	P3-Ru1-P4	71.33 (3)
Cl1-Ru1-P2	163.69 (3)	Cl1-Ru1-O1	81.41 (9)
P1-Ru1-P2	71.08 (3)	P1-Ru1-O1	88.92 (9)
Cl1-Ru1-P3	93.70 (3)	P2-Ru1-O1	95.83 (9)
P1-Ru1-P3	172.00 (4)	P3-Ru1-O1	96.74 (9)
P2-Ru1-P3	102.60 (3)	P4-Ru1-O1	166.73 (9)
Cl1-Ru1-P4	93.34 (4)	P1-C49-P2	94.24 (15)
P1-Ru1-P4	103.56 (4)	P3-C50-P4	96.68 (18)
P2-Ru1-P4	92.54 (3)		

The crystallographic asymmetric unit consists of one $[RuCl(OH_2)(PPh_2CH_2PPh_2)_2]^+$ cation, one PF_6^- anion and methanol molecules of solvation at two general locations. One methanol molecule (atoms O3 and C52) appears to have an occupancy of less than 1.0. When it is present, the other methanol is sited at O2 and C51, and the hexafluorophosphate F atoms are at F1-F6; when it is absent, the other methanol is at O21 and C511, and the hexafluorophosphate F atoms occupy sites F1, F21, F31, F4, F51 and F61, corresponding to a rotation about the F1-P5-F4 axis. The relative occupancies of these alternatives were refined, the final values being two 0.730 (5):0.270 (5). The O2-C51 and O21-C511 distances were restrained to be equal. The largest peaks in the final difference electron density map are located near the Ru atom. The space group is non-centrosymmetric but contains glide planes, giving a racemic structure. H atoms attached to C atoms of the cation were included at idealized positions and allowed to ride on the atoms to which they are bonded, with C-H = 1.0 Å and $U_{iso}(H) = 1.2U_{eq}(C)$. H atoms of the aqua ligand and of the methanol molecules were not located.

Data collection: COLLECT (Nonius, 1997); cell refinement: DENZO and SCALEPACK (Otwinowski & Minor, 1997); data



0.079 and 0.197 (Carruthers & Watkin, 1979) $(\Delta/\sigma)_{max} = 0.019$ $\Delta\rho_{max} = 0.69 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -1.37 \text{ e} \text{ Å}^{-3}$ Absolute structure: Flack (1983), 5819 Friedel pairs Flack parameter = -0.02 (2)

Figure 1

C35

C36

A view of the cation of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms have been omitted.

reduction: *DENZO* and *SCALEPACK*; program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Watkin *et al.*, 2001); molecular graphics: *ORTEP*II (Johnson, 1976) in *TEXSAN* (MSC, 1992–1997); software used to prepare material for publication: *CRYSTALS*.

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